# ALLOYS, CARBON & ITS VARIOUS FORMS, HYDROGEN & COMPOUNDS, SULPHUR, NITROGEN, HALOGEN & INERT GASES BASED GENERAL SCIENCE MCQ PRACTICE QUESTIONS AND ANSWERS PDF WITH EXPLANATION

For All Competitive SSC, Bank, IBPS, UPSC, Railway, IT & Other Govt. Exams

Created By Careericons Team

<b>Q1.</b> Buckminster	Ful	lerene	is
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- a) None of the above
- b) A form of carbon compound of clusters of 60 carbon atoms bond together in polyhedral structure composed of pentagons or hexagons
- c) A polymer of fluorine
- d) An isotope of carbon heavier than C14

### Q2. Chemically Dry Ice is

- a) Solid carbon dioxide
- b) Solid sulphur
- c) Ice made from distilled water
- d) Mixture of ice and common salt
- **Q3.** Percentage amount (by volume) of which one of the following gas is lowest in the atmosphere?
- a) Carbon dioxide
- b) Nitrogen
- c) Argon
- d) Oxygen
- Q4. Which lubricant is used for heavy machines?
- a) Silicon oil

b) Bauxitec) Phosphorusd) Graphite

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Q5. Consider these statements and choose the right answer from counterfeit.

## Statement (A):

Normally it has been seen that iron goods covered with a brown powder called rust when they are unsheltered in the atmosphere.

## Statement (R):

Rust or the brown powder is the result of deposition Tannin. Code

- a) (A) is false, but (R) is trues
- b) (A) is true, but (R) is false
- c) Both (A) and (R) are true and (R) is the correct explanation of (A)
- d) Both (A) and (R) are true, but (R) is not the correct explanation of (A)
- Q6. The important metal used with iron to make it rust free
- a) Tin
- b) Chromium
- c) Aluminium
- d) Carbon
- Q7. Which one of the following metals does not form amalgam?
- a) Iron
- b) Magnesium
- c) Zinc

d) Copper
Q8. Burning of Hydrogen produces
a) Water
b) Soil
c) Oxygen
d) Ash
Q9. What is the shape of sulphur hexafluoride molecule?
a) Octahedral
b) Planar
c) Trigonal pyramid
d) Tetrahedral
Q10. Which metal exists as a common component in Brass, Bronze and German Silver?
a) Zinc
b) Tin
c) Antimony
d) Copper
Q11. Which gas is not in the gaseous cycle?
a) \$O_2\$
b) Carbon
c) \$N_2\$
d) \$H_2\$

# Q12. Consider the following statements and select the correct answer from given codes below.

- Water becomes harder due to the presence of Calcium Sulphate and it is not usable.
- Diamond is harder than copper and iron.
- Oxygen is the main component of air.
- Nitrogen is used in the manufacture of vegetable ghee.

#### Code

- a) 2 and 4
- b) 1 and 2
- c) 3 and 4
- d) 1 and 3

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# Q13. Large quantities of drinking water is prepared from impure water by

- a) Decantation
- b) Ion-exchange
- c) Desalination
- d) Distillation

# Q14. Water is a good solvent of ionic salts because

- a) It has no colour
- b) It has a high specific heat
- c) It has a high boiling point
- d) It has a high dipole moment

# Q15. By which process, Alum cleans the muddy water?

a) Dialysis b) Coagulation c) Absorption d) Adsorption Read More chemistry gk quiz Question and Answes » Answers to the above questions: Q1. Answer: (b) Q2. Answer: (a) Q3. Answer: (a) 5000+ GENERAL SCIENCE MCQ TOPIC WISE MCQ QUESTION BANK WITH SOLVED **ANSWERS & FREE PDF** MEASUREMENT AND SI UNITS MEASURING DEVICES AND SCALES MECHANICS (MASS, FORCE, ACCELERATION, WORK, ENERGY, MOMENTUM, VELOCITY) FORCE OF GRAVITY, SOLIDS AND FLUIDS MOTION UNDER GRAVITY PHYSICAL PROPERTIES OF MATERIALS LIGHT, HEAT, THERMODYNAMICS, WAVE MOTION, SOUND, OPTICS & OSCILLATIONS NUCLEAR PHYSICS AND CONDUCTIVITY ELECTRIC CURRENT AND MAGNETISM MODERN AND SPACE PHYSICS, SOURCES OF ENERGY

GENERAL CONCEPTS OF CHEMISTRY NATURE OF MATTER AND COMPOSITION

ATOMIC STRUCTURE MOLECULES AND NUCLEAR CHEMISTRY

# CLASSIFICATION OF ELEMENTS, CHEMICAL BONDING, PERIODICITY IN PROPERTIES & PHYSICAL TRANSFORMATION

METALS, NON METALS, MINERALS, ORES, PROPERTIES, USES

ALLOYS, CARBON & ITS VARIOUS FORMS, HYDROGEN & COMPOUNDS, SULPHUR, NITROGEN, HALOGEN & INERT GASES

ACIDS, BASES, SALTS AND METALS

HYDROCARBON, ALCOHOL, POLYMER, ORGANIC ACID AND EXPLOSIVE SUBSTANCE FUEL, FOOD PRESERVATION, NUTRITION, MEDICINE, DETERGENTS AND FERTILIZERS INORGANIC CHEMISTRY ORGANIC CHEMISTRY ENVIRONMENTAL CHEMISTRY SOME IMPORTANT MAN MADE MATERIALS SUB DIVISIONS OF BIOLOGY GENETICS, EVOLUTION OF LIFE & ECOLOGICAL BIODIVERSITY TAXONOMY HEREDITY

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GENETIC ENGINEERING AND BIO TECHNOLOGY

Q4. Answer: (d)

Q5. Answer: (b)

Q6. Answer: (b)
Q7. Answer: (a)
Q8. Answer: (a)
Q9. Answer: (a)
Q10. Answer: (d)
Q11. Answer: (d)
Q12. Answer: (b)
Q13. Answer: (c)
Q14. Answer: (d)
Q15. Answer: (b)

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